Name $\qquad$

How many carbon atoms are in $\mathrm{C}_{8} \mathrm{H}_{18}$ ? _ 8 _ How many hydrogen atoms? _18 $\qquad$

Write the chemical formula for a compound with 2 hydrogen atoms and one oxygen atom.

$$
\mathrm{H}_{2} \mathrm{O}
$$

Indicate the number of each type of atom in the following compound.

$$
\mathrm{Ca}_{3}\left(\mathrm{PO}_{4}\right)_{2}
$$

$\mathrm{Ca} 3 \quad \mathrm{P}$ 3_2 $\quad 0$ 8

Indicate the number of each type of atom in vitamin B-12.

|  | $\mathrm{C}_{63} \mathrm{H}_{88} \mathrm{CON}_{14} \mathrm{O}_{14} \mathrm{P}$ |  |
| :---: | :---: | :---: |
| C 63 | H $\quad 88$ | Co |
| N__14 | $0 \_14$ | P __1 |

Write the formula for a compound that has 17 carbon atoms, 20 hydrogen atoms, 4 nitrogen atoms, and 6 oxygen atoms.

$$
\mathrm{C}_{17} \mathrm{H}_{20} \mathrm{~N}_{4} \mathrm{O}_{6}
$$

A _compound $\qquad$ is a pure substance that can be broken into simpler substances by chemical means.

