

Percentage and Parts per Million

1. There are a total of 58 students in a class. The last exam had 35% receive an A, 24% receive a B, and the remaining students received a C. How many C's were received?

$$100\% - (35\% + 24\%) = 41\% \text{ received a C.}$$

$$41/100 = 0.41$$

$$0.41 \times 58 = \mathbf{24 \text{ students received a C.}}$$

2. A gaming club has 35% more gamers in Level 1 than in Level 2. There are a total of 145 gamers in Level 1. How many gamers are in Level 2?

$$\text{The percentage of gamers in level 1 is } 100\% + 35\% = 135\%$$

$$135/100 = 1.35 \text{ We can use } 1.35x \text{ to represent the number in}$$

$$\text{Level 1. So, } 1.35x = 145. \text{ Solve for } x. \text{ } x = \mathbf{107 \text{ gamers in Level 2.}}$$

3. A compound, NaHCO_3 (baking soda) has a mass of 84.0 g. The percent by mass of Na is 27.4%, C is 14.3%, and H is 1.20%. What is the percentage of oxygen in the compound?

$$\% \text{ Oxygen} = 100\% - (27.4\% + 14.3\%) = \mathbf{58.3\%}$$

4. A large creek is located by a steel mill. A 2622 g sample of water was found to contain 0.0075 g of Cr^{6+} ion. Chromium (VI) ion is a carcinogen. Calculate the following:

percentage of Cr^{6+}

$$\frac{0.0075 \text{ g}}{2622 \text{ g}} \times 100 = \mathbf{0.00029\%}$$

ppm of Cr^{6+}

$$\frac{0.0075 \text{ g}}{2622 \text{ g}} \times 10^6 = \mathbf{2.9 \text{ ppm}}$$

ppb of Cr^{6+}

$$\frac{0.0075 \text{ g}}{2622 \text{ g}} \times 10^9 = \mathbf{2.9 \times 10^3 \text{ ppb}}$$