

Significant Figures

Name _____

Significant figures are the number of meaningful digits used to express a value. The number of digits include the digits that are known with certainty plus one estimated digit which is usually unknown by ± 1 .

- Leading zeros are not significant.
- Zeros at the end of a number are not significant unless they are to the right of the decimal point.

Indicate the number of significant figures in each measurement.

240 mL 2

4.53 g 3

2.5000×10^3 mL 5

0.025 g/mL 2

0.0400 mg 3

0 °F 1

852000 kg 3

860. g 3

24.000 mi. 5

0.00 mL 3

0.0700 mg 3

24000006 kg 8

For **addition** and **subtraction**, just look at the number of digits to the right of the decimal point.

For **multiplication** and **division**, look at the number of significant figures in each measurement.

Do the following calculations, and report answers to the correct number of significant figures

1.) $256.44 \text{ g} / 208 \text{ mL}$

$$1.23 \frac{\text{g}}{\text{mL}}$$

2.) $\frac{640 \text{ mL} - 254 \text{ mL}}{25.22 \text{ g} + 86.1 \text{ g}}$

$$3.47 \frac{\text{mL}}{\text{g}}$$

3.) $\frac{4.7 \times (8.62 - 3.400)}{6.0 \times 0.25}$

$$16$$

4.) $\frac{0.4266 \text{ g} + 1.66 \text{ g}}{56.240 \text{ mL} - 32.62 \text{ mL}}$

$$0.883 \frac{\text{g}}{\text{mL}}$$

5.) $\frac{925.6002 + 38.7340}{2.6200}$

$$368.07$$

6.) $0.886 \times (32.40 \text{ mg} - 6.257 \text{ mg})$

$$23.2 \text{ mg}$$